

# GA<sub>4</sub> Does Not Require Conversion into GA<sub>1</sub> to Delay Senescence of *Alstroemeria hybrida* Leaves

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Abstract. The biological activity and metabolism of applied GA<sub>1</sub> and GA<sub>4</sub> were studied in leaves of alstroemeria (Alstroemeria hybrida). It appeared that  $GA_4$  was 2 orders of magnitude more active in delaying leaf senescence than GA<sub>1</sub>. GA<sub>3</sub>-13-OMe, a GA analog that cannot be hydroxylated on the 13-C position, also retarded chlorophyll loss, although less efficiently. Tritiated and deuterated GA<sub>1</sub>, GA<sub>4</sub>, and GA<sub>9</sub> were applied to leaves, and their metabolites were analyzed. According to high performance liquid chromatography and gas chromatography-mass spectrometry analyses, GA<sub>9</sub> was converted into GA<sub>4</sub> and GA<sub>34</sub>, and GA<sub>4</sub> was converted into GA<sub>34</sub> and more polar components. No evidence was found for the conversion of both GA<sub>9</sub> and GA<sub>4</sub> into GA<sub>1</sub>, even at the relatively high concentrations that were taken up by the leaf. The results strongly suggest that  $GA_4$  is recognized directly by a receptor involved in regulation of leaf senescence in alstroemeria.

Key Words. Alstroemeria—Gibberellins—Metabolism—Senescence

Gibberellins (GAs) play an important role in the control of growth and development of plants. It has been hypothesized that different types of GAs are involved in different processes; vegetative and generative processes might be regulated by specific GAs (Sponsel 1995).

There is considerable evidence that GA<sub>1</sub> has biological

activity per se especially in regulating seed germination and shoot and internode elongation (Graebe 1987, Sponsel 1995). Next to GA<sub>1</sub>, a number of studies also indicate GA<sub>4</sub> as biologically active in regulating stem elongation. Metabolism of GA<sub>4</sub> has been studied in various systems, and in seedlings of Phaseolus (Turnbull and Crozier 1989), Zea, Oryza, and Arabidopsis (Kobayashi et al. 1991), GA<sub>4</sub> can be metabolized to GA<sub>1</sub>. Thus, the biological activity of applied GA<sub>4</sub> may be the result of its conversion to GA<sub>1</sub>. However, GA<sub>4</sub> has been suggested to be active per se for stimulation of stem elongation in Cucumus sativus (Nakayama et al. 1991), Arabidopsis thaliana (Talon et al. 1990), and Tulipa gesneriana (Rebers et al. 1995) based on a higher endogenous  $GA_4$ concentration compared with GA<sub>1</sub>. In addition, GA<sub>34</sub> is a common metabolite of GA<sub>4</sub> in vegetative tissues of Phaseolus coccineus (Turnbull and Crozier 1989). These references suggest the existence of the non-early-3,13hydroxylation pathway yielding GA<sub>4</sub> as intrinsically active GA.

Limited information is available as to whether  $GA_1$  is also the most active GA in regulating other plant developmental processes such as senescence. Alstroemeria leaves offer a good model system to study senescence closely related to the in situ situation. It has the advantage of being highly sensitive to GAs, and the time course of the dark-induced senescence of detached leaf tips is very comparable with that of attached leaves (Jordi et al. 1993). In previous studies with alstroemeria, endogenous GA concentrations have been related to leaf senescence (Kappers et al. 1997). It has been shown that several GAs of both the 13-hydroxylation and the non-13-hydroxylation pathways were present in these leaves. In fresh leaves, GA<sub>4</sub> and GA<sub>1</sub> were present in comparable concentrations (approximately  $0.7 \text{ pmol} (\text{g FW})^{-1}$ ). GAs of the non-13-hydroxylation pathway changed in concentration because of senescence-inducing treatment (darkness), whereas 13-hydroxylated GAs did not. GA<sub>4</sub>

**Abbreviations:** GA(s), gibberellin(s); FW, fresh weight; HPLC, high performance liquid chromatography; GC-MS, gas chromatographymass spectrometry; SIM, selected ion monitoring.

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especially was influenced; after 3 days of dark treatment no  $GA_4$  could be detected. Under low red light conditions, which delayed leaf senescence, the concentration of  $GA_4$  remained at a much higher level (Kappers et al. 1998). Apparently, the decrease in  $GA_4$  is a very early response of the senescence process. Exogenously applied  $GA_4$  was found to be active at a concentration that was several orders of magnitude lower than that of  $GA_1$ (Jordi et al. 1995, Kappers et al. 1998). This observed difference might be explained in terms of uptake of GAs by the tissue or in terms of metabolism of active GAs. Additionally, applied  $GA_4$  could function as a precursor (slow release form) for  $GA_1$ , whereas exogenously applied  $GA_1$  is inactivated rapidly.

The focal point of this study with alstroemeria leaves is to compare the biological activity and metabolism of exogenously added  $GA_1$  and  $GA_4$  to elucidate whether the involved putative receptor has the ability to recognize  $GA_4$ . Dose-response experiments were performed with  $GA_1$ ,  $GA_4$ , and  $GA_3$ -13-OMe, a  $GA_3$ -analog that is unlikely to be converted back into  $GA_3$ , as in  $GA_1$  hydroxylated at 13-C. In metabolic studies with <sup>2</sup>H<sub>2</sub>- and <sup>3</sup>H<sub>3</sub>labeled  $GA_1$ ,  $GA_4$ , and  $GA_9$ , GA metabolites were determined with HPLC and GC-MS. Our results suggest that in delaying senescence in alstroemeria leaves  $GA_4$  is biologically active per se and is probably recognized by the specific receptor involved.

#### **Materials and Methods**

#### Chemicals

GA<sub>1</sub>, GA<sub>4</sub>, GA<sub>9</sub> (purity > 99%),  $[17^{-2}H_2]GA_1$ ,  $[17^{-2}H_2]GA_4$ , and  $[17^{-2}H_2]GA_9$  were obtained from Prof. L. N. Mander (Canberra, Australia); GA<sub>3</sub>-13-OMe was a gift from Dr. M. H. Beale (Bristol, UK). GC-MS control experiments demonstrated that GA<sub>3</sub>-13-OMe contains  $\leq 0.01\%$ of other GAs such as GA<sub>1</sub> and GA<sub>4</sub>.  $[1,2^{-3}H_2]GA_1$  and  $[1,2^{-3}H_2]GA_4$ were obtained from Amersham (Buckinghamshire, UK) and  $[2,3^{-3}H_2]GA_9$  from Dr. A. Crozier (Glasgow, UK). The specific activity of tritiated GAs was about 50,000 Ci mol<sup>-1</sup>. Stock solutions in ethanol were prepared at  $10^{-2}$  M for the various GAs. In all experiments the final ethanol concentration was  $\leq 1\%$  v/v. Control experiments demonstrated that this ethanol concentration did not affect the rate of chlorophyll loss (data not shown).

#### Plant Material and Dose-Response Experiments

The five uppermost leaves of *Alstroemeria hybrida* cv. Cinderella, positioned in a whorl directly below the flowers and therefore of the same physiological age, were used for the experiments. Leaf tips were excised with a sharp knife. The weight of the leaf tips was determined, and the leaf tips (FW approximately 500 mg, leaf area approximately 7 cm<sup>2</sup>) were incubated in various concentrations of GAs (total volume 3 mL) in darkness. Logistic dose-response curves were calculated from the individual data points, using a nonlinear regression method described by the formula ( $R = R_{min} + (R_{max} - R_{min})/(1 + ([GA]/[GA]_{50})^p)$ , where *R* is the response,  $R_{min}$  is the response in the absence of exogenous GA,  $R_{max}$  is the maximum response, [GA] is the applied

concentration,  $[GA]_{50}$  is the concentration required for half-maximum response, and *p* is the Hill coefficient representing the abruptness of the curve (Weyers et al. 1987).

#### Chlorophyll Analysis

The chlorophyll (a + b) content of the leaves was calculated using the molar extinction coefficient in dimethylformamide (Inskeep and Bloom 1985). Data are presented as the percentage of the initial amount of chlorophyll in order to correct for small differences between experiments.

## Uptake and Metabolism of Deuterated and Tritiated GAs

Tritiated GAs were first purified by HPLC before feeding. Each leaf was placed in 3 mL of a  $10^{-7}$  M GA solution enriched with 50,000 dpm of tritiated labeled GA. In experiments in which combinations of deuterated and tritiated GAs were fed, GAs were diluted in 0.1 M citrate phosphate buffer, pH 5.0. Leaf tips were incubated in darkness at 20°C. After incubation, leaves were frozen in liquid N<sub>2</sub> and stored at -80°C until extraction. Experiments were performed in duplicate with four replications in each.

#### Extraction and Purification of GAs

Leaves were homogenized in ice-cold 100% MeOH, 5 mL g FW<sup>-1</sup>, containing 0.1% ascorbic acid. When leaves were fed both deuterated and tritiated GAs, three leaves were combined, making one sample. After extraction for 2 h at 4°C and continuous shaking, the extract was centrifuged (5 min at 5,000  $\times$ g), and the pellet was washed twice with 80% MeOH. The pooled MeOH extracts were evaporated to the water phase under reduced pressure at 35°C. The aqueous phase ( $\pm 2$  mL) was adjusted to pH 2.5 with 1 N HCl and partitioned against ethyl acetate (2 mL three times). The combined ethyl acetate fractions were evaporated to dryness, dissolved in 2 mL of water, and adjusted to pH 8.0 with 1 N NaOH. The extract was then purified further through a QAE-Sephadex A-25 (Pharmacia, Uppsala, Sweden) column (4 cm  $\times$  0.25 cm, inner diameter) equilibrated with sodium formate (1% w/v) and fed directly through a preequilibrated  $C_{18}$  column (Waters, Millipore Corporation, Milford, MA, USA). GAs were eluted with 80% MeOH and evaporated to dryness under reduced pressure before HPLC and GC-MS analysis.

#### HPLC and GC-MS

The HPLC system and operating conditions are described in Kappers et al. (1997). The column was eluted at a flow rate of 1 mL min<sup>-1</sup> with 30% methanol for 15 min followed by a linear gradient to 75% methanol over 35 min and subsequently to 90% methanol over 5 min (solvents contained 0.01% acetic acid).

To analyze metabolic conversion products, successive 0.5-mL fractions were collected, and the radioactivity in an aliquot of each fraction was assayed by liquid scintillation counting. In experiments with both tritiated and deuterated labeling, fractions containing radioactivity were analyzed by GC-SIM as described by Kappers et al. (1997). For the detection of  $[^{2}H_{2}]GAs$ , their characteristic ions were monitored together with the characteristic ion of the endogenous GA. The labeled metabolites were identified based on co-chromatography with the en-



**Fig. 1.** Effect of various concentrations of  $GA_1$  (•),  $GA_4$  (•), and  $GA_3$ -13-OMe (•) on chlorophyll retention in alstroemeria leaf tips placed in darkness for 10 days. Data are the means of two experiments with five replications each. *Vertical bars* represent the standard error. The *inset* shows the doses needed for the half-maximum response of the tested gibberellins ([GA]<sub>50</sub>).

dogenous GAs, which had been identified previously using Kováts retention indices and full-scan mass spectra compared with those of authentic standards (Kappers et al. 1997) except for GA<sub>8</sub> and GA<sub>34</sub> catabolites, which were identified tentatively by comparison with data from Gaskin and MacMillan (1991).

#### **Results and Discussion**

#### GAs Delay Senescence

There was no significant loss of chlorophyll during the first 5 days in all treatments (data not shown). Thereafter, chlorophyll decreased rapidly from day 6 onward in leaves that were placed in darkness, and leaves were uniformly yellow after 10 days. Application of a number of GAs delayed the loss of chlorophyll (Jordi et al. 1995). We determined the effect of the various treatments after 10 days of incubation, when dark-treated leaves without GA retained only 20% of the initial amount of chlorophyll.

Fig. 1 shows the effects of various concentrations of  $GA_1$ ,  $GA_4$ , and  $GA_3$ -13-OMe chlorophyll loss in leaf tips of alstroemeria placed in darkness for 10 days. Both  $GA_1$  and  $GA_4$  showed a concentration-dependent effect on the retention of chlorophyll with a maximum effect of about 80% of the initial amount of chlorophyll left (Fig. 1). However, the efficacy of  $GA_4$  is 2 orders of magnitude higher than  $GA_1$  as can be seen from the calculated concentrations needed for the half-maximum effect ( $[GA]_{50}$ ) (2.1 and 220 nM, respectively). These data indicate that the receptor involved will recognize exogenously applied  $GA_4$ . However, a number of arguments can be proposed which argue against the simplicity of this conclusion. Therefore, we investigated the relation between  $GA_1$  and  $GA_4$  further.

**Table 1.** Uptake of  $[{}^{3}H_{2}]GA_{1}$  and  $[{}^{3}H_{2}]GA_{4}$  by alstroemeria leaf tips during 48 h in darkness.

Incubation in dark (h)	[ <sup>3</sup> H <sub>2</sub> ]GA <sub>1</sub> (dpm)	% of applied label	[ <sup>3</sup> H <sub>2</sub> ]GA <sub>4</sub> (dpm)	% of applied label
5	123	0.2	178	0.4
10	355	0.7	366	0.7
24	560	1.1	467	0.9
48	802	1.6	970	1.9

### Uptake of $[{}^{3}H]GA_{1}$ and $[{}^{3}H]GA_{4}$

First, differences in the uptake of  $GA_1$  and  $GA_4$  might explain the differences found in efficacy. Leaves were incubated with a  $10^{-7}$  M GA solution enriched with tritium-labeled GA. Leaves were shown to have a comparable uptake during the experimental period of 48 h (Table 1). Thus, differences in GA uptake cannot explain the differences in the efficacy of  $GA_1$  and  $GA_4$ .

#### 13-Hydroxylation Necessity

It is thought that active GAs have two characteristics in common: a 10-C–19-C lactone bridge below the plane of the A-ring and a hydroxyl group at the 3-C position (Pearce et al. 1994). According to Stoddart (1986) hydroxylation of the 13-C position enhances efficacy further. The presence of hydroxyl groups cannot simply explain the activity in alstroemeria leaves because the less hydroxylated  $GA_4$  was much more active than  $GA_1$ .

To find out whether conversion of GA<sub>4</sub> into GA<sub>1</sub> and hence a 13-OH group is a prerequisite for activity we used a GA analog. This modified GA has an O-CH<sub>3</sub> group on the 13-C position and is unlikely to be converted back to its basis GA (Dr. M. H. Beale, personal communication). From previous work it is known that an order of efficacy exists in GA acitivity from  $GA_4 = GA_7$  $> GA_9 > GA_1 > GA_3$  (Jordi et al. 1995). We used  $GA_3$ -13-OMe to test whether a 13-OH group is necessary for biological activity. From the results in Fig. 1 it is clear that GA<sub>3</sub>-13-OMe showed a concentration-dependent effect on the delay of chlorophyll loss. No GA<sub>1</sub>, GA<sub>3</sub>, or GA<sub>4</sub> was detected in an aliquot of 100 ng of GA<sub>3</sub>-13-OMe analyzed by GC-MS. Although the calculated concentration needed for the half-maximum response is 310 nM for GA<sub>3</sub>-13-OMe compared with 2.1 nM for GA<sub>4</sub>, it is clear that GA<sub>3</sub>-13-OMe delays leaf senescence. The observed difference in maximum activity could be partly a result of the difference found between the maximum activity of  $GA_4$  and  $GA_1$ . We conclude that hydroxylation on the C-13 position seems not essential for delaying chlorophyll loss.

#### Metabolism of $GA_1$ , $GA_4$ , and $GA_9$

Our results show that uptake of GAs cannot explain the differences found in efficacy. Furthermore, 13-hydrox-ylation of GAs is not a prerequisite for biological activity

	Deuterated GAs <sup>a</sup>		Relative neak area ratio		
GA fed to the leaves	KRI 2,689	GA <sub>n</sub>	$[m/z + 2]/[m/z]^a$		Enrichment
GA <sub>1</sub> standard			0.150	(508/506)	
GA <sub>4</sub> standard	2,522		0.053	(286/284)	
GA <sub>8</sub> standard	2,837		0.254	(596/594)	
GA <sub>9</sub> standard	2,319		0.034	(300/298)	
GA <sub>34</sub> standard	2,675		0.169	(508/506)	
GA <sub>8</sub> -catabolite <sup>b</sup>	2,723		0.182	(536/534)	
GA <sub>34</sub> -catabolite <sup>b</sup>	2,569		0.294	(373/371)	
[ <sup>2</sup> H <sub>2</sub> ]GA <sub>1</sub> fed	2,689	$GA_1$	$2.257\pm0.12$	(508/506)	15.05
	2,522	$GA_4$	$0.059 \pm 0.003$	(286/284)	1.11
	2,837	GA <sub>8</sub>	$1.056 \pm 1.89$	(596/594)	4.22
	2,319	$GA_9$	$0.039 \pm 0.004$	(300/298)	1.15
	2,675	GA <sub>34</sub>	$0.153 \pm 0.009$	(508/506)	0.91
	2,723	GA <sub>8</sub> -cat	$0.319\pm0.18$	(536/534)	1.75
	2,569	GA <sub>34</sub> -cat	$0.291 \pm 0.06$	(373/371)	0.99
[ <sup>2</sup> H <sub>2</sub> ]GA <sub>4</sub> fed	2,689	$GA_1$	$0.138\pm0.01$	(508/506)	0.92
	2,522	$GA_4$	$1.864 \pm 0.09$	(286/284)	35.17
	2,837	$GA_8$	$0.272 \pm 0.04$	(596/594)	1.07
	2,319	$GA_9$	$0.040 \pm 0.002$	(300/298)	1.18
	2,675	GA <sub>34</sub>	$17.601 \pm 1.27$	(508/506)	104.15
	2,723	GA8-cat	$0.178 \pm 0.04$	(536/534)	0.98
	2,569	GA <sub>34</sub> -cat	$0.641 \pm 0.12$	(373/371)	2.18
[ <sup>2</sup> H <sub>2</sub> ]GA <sub>9</sub> fed	2,689	$GA_1$	$0.162\pm0.005$	(508/506)	1.08
	2,522	$GA_4$	$1.532\pm0.08$	(286/284)	28.91
	2,837	GA <sub>8</sub>	$0.292\pm0.02$	(596/594)	1.15
	2,319	$GA_9$	$3.029 \pm 0.13$	(300/298)	89.08
	2,675	GA <sub>34</sub>	$16.431\pm0.88$	(508/506)	97.22
	2,723	GA <sub>8</sub> -cat	$0.180\pm0.05$	(536/534)	0.99
	2,569	GA <sub>34</sub> -cat	$0.451\pm0.04$	(373/371)	1.53

**Table 2.** GC-SIM data on metabolites from  $[^{2}H_{2}]GA_{1}$ ,  $[^{2}H_{2}]GA_{4}$ , and  $[^{2}H_{2}]GA_{9}$  incubation of alstroemeria leaves for 48 h. Data are means of four independent experiments  $\pm$  S.D. Enrichment is calculated by ([m/z + 2]/[m/z] of sample)/([m/z + 2]/[m/z] of standard).

<sup>a</sup> Ions representing [m/z + 2] and [m/z] for the different GAs are given in parentheses.

<sup>b</sup> Obtained by interpolation of data of Gaskin and MacMillan (1991).

in the delay of leaf senescence. We have studied further the metabolism of deuterated  $GA_4$  and  $GA_1$  to elucidate the possible conversion pathways in vivo of these GAs.

In alstroemeria leaves, endogenous amounts of GA<sub>1</sub> and GA4 in mature (fresh) leaves are almost identical:  $0.66 \pm 0.05 \text{ pmol} (\text{g FW})^{-1} \text{ and } 0.72 \pm 0.02 \text{ pmol} (\text{g FW})^{-1}$ FW)<sup>-1</sup>, respectively (Kappers et al. 1998). To get insight in their metabolism, we followed the fate of deuterated and tritiated labeled compounds. Although to date it is nearly impossible to estimate exact turnover rates, we can compare relative turnover rates of both GAs. To ensure a high uptake of deuterated GAs into the leaf, the GAs were dissolved in 0.1 M citrate phosphate buffer at pH 5.0. Previous experiments showed that uptake was maximal at this pH (data not shown). The amount of deuterated GA taken up by the leaf was calculated from the tritiated GAs. Assuming no discrimination in the uptake of deuterated and tritiated GA, the uptake of deuterated GA was estimated to be 65 pmol of GA<sub>1</sub>, 59 pmol of GA<sub>4</sub>, and 182 pmol of GA<sub>9</sub>/(qFW)<sup>-1</sup>, resulting in an increase of the endogenous concentration of GA1, GA4,

and GA<sub>9</sub> with a factor 98, 82, and 152, respectively. When leaves were incubated with  $[^{2}H_{2}]GA_{4}$  the ratio m/z286/284, an estimation of the ratio deuterated vs nondeuterated GA<sub>4</sub>, found after 48 h of labeling was 35 times the ratio found in nonlabeled leaves, which confirmed an increased endogenous concentration of GA<sub>4</sub> (Table 2). The discrepancy between the total uptake of 82 times the endogenous concentration and the enrichment of the ratio between deuterated and endogenous GA (35 times) is caused by the metabolism of the applied GA. The increased ratios of 508/506 and 373/371 found at the retention times of GA34 and GA34 catabolite suggested the presence of deuterated GA<sub>34</sub> and possibly GA<sub>34</sub> catabolite, demonstrating the conversion of GA<sub>4</sub> into GA<sub>34</sub>. This step involves a 2 $\beta$ -hydroxylation, known to inactivate GAs (Graebe 1987). GA34 was found in alstroemeria leaves at a low concentration of 0.04 pmol  $(g FW)^{-1}$  (Kappers et al. 1997). After incubation with  $GA_4$ , endogenous  $GA_1$  was detected, but the m/z 508/506 ratio observed corresponded to the natural abundance of the isotope. This indicated that even at a very high endogenous  $GA_4$  concentration no significant extra  $GA_1$  was formed (i.e. less than 3%). To exclude as much as possible experimental errors in demonstrating  $GA_4$  metabolism, we also incubated leaves with labeled  $GA_9$  (Table 2). It can be seen that the ratios of  $GA_9$ ,  $GA_4$ , and  $GA_{34}$  differed from the endogenous values. Also in this case no increase in endogenous  $GA_1$  could be found. Incubation of leaves with deuterated  $GA_1$  led to increased ratios for  $GA_1$  and  $GA_8$  and possibly  $GA_8$  catabolite (Table 2).

Although the estimated concentration of the biologically active hormones was the same for  $GA_1$  and  $GA_4$ , the turnover rate, that is, the number of molecules/time unit formed and subsequently turned over, seemed to be different. From the calculated enrichment (Table 2) it can be concluded that the non-13-hydroxylated  $GA_{34}$  and GA<sub>34</sub> metabolite were formed relatively more than their corresponding 13-hydroxylated counterparts, GA<sub>8</sub> and  $GA_8$  catabolite. Although the uptake of both  $GA_1$  and  $GA_4$  is comparable, the enrichment of the leaf with  $GA_1$ is lower after 48 h. This indicates a higher turnover rate for  $GA_1$  than for  $GA_4$ . However, it is obvious that even at high concentrations of GA4 in the leaf no conversion into GA1 occurs. Combined with the greater biological efficacy of applied GA<sub>4</sub> toward delay of senescence this makes it very likely that the non-13-hydroxylation pathway yielding GA<sub>4</sub> is a part of early processes regulating senescence in alstroemeria leaves. As no conversion into  $GA_1$  is needed,  $GA_4$  will be recognized directly by the receptor involved in the regulation of leaf senescence.

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